

Hexaaquamagnesium(II) bis[4-(2-hydroxybenzylideneamino)benzenesulfonate]

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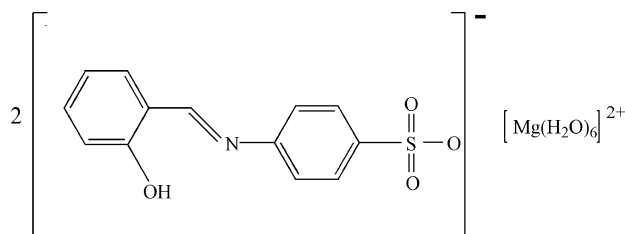
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Key indicators: single-crystal X-ray study; $T = 291$ K; mean $\sigma(\text{C}-\text{C}) = 0.006$ Å; R factor = 0.064; wR factor = 0.146; data-to-parameter ratio = 14.8.

In the crystal structure of the title compound, $[\text{Mg}(\text{H}_2\text{O})_6](\text{C}_{13}\text{H}_{10}\text{NO}_4\text{S})_2$, the packing is stabilized by $\text{O}_{\text{water}}-\text{H}\cdots\text{O}_{\text{anion}}$ hydrogen bonds. An intramolecular $\text{O}-\text{H}\cdots\text{N}$ bond occurs in the anion. The Mg atom has site symmetry $\bar{1}$.

Related literature

 For related literature, see: Tai *et al.* (2003).


Experimental

Crystal data

 $[\text{Mg}(\text{H}_2\text{O})_6](\text{C}_{13}\text{H}_{10}\text{NO}_4\text{S})_2$
 $M_r = 684.97$

 Monoclinic, $P2_1/n$
 $a = 6.2997$ (17) Å

 $b = 35.313$ (9) Å

 $c = 6.9561$ (19) Å

 $\beta = 90.75$ (1)°

 $V = 1547.3$ (7) Å³
 $Z = 2$

 Mo $K\alpha$ radiation

 $\mu = 0.26$ mm⁻¹
 $T = 291$ (2) K

 $0.30 \times 0.24 \times 0.22$ mm

Data collection

Bruker SMART APEX CCD diffractometer

 Absorption correction: multi-scan (*SADABS*; Bruker, 2000)

 $T_{\text{min}} = 0.93$, $T_{\text{max}} = 0.94$

8390 measured reflections

3041 independent reflections

 2173 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.044$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.064$
 $wR(F^2) = 0.146$
 $S = 1.00$

3041 reflections

205 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.25$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.46$ e Å⁻³
Table 1

Selected bond lengths (Å).

| | | | |
|--------|-----------|--------|-----------|
| Mg1—O5 | 2.031 (2) | Mg1—O7 | 2.071 (2) |
| Mg1—O6 | 2.055 (2) | | |

Table 2

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|-----------------------------------|-------|-------------|-------------|---------------|
| O1—H1C \cdots N1 | 0.85 | 1.78 | 2.574 (5) | 154 |
| O5—H5A \cdots O4 ⁱ | 0.96 | 2.03 | 2.732 (3) | 129 |
| O5—H5B \cdots O3 | 0.96 | 1.87 | 2.768 (3) | 154 |
| O6—H6B \cdots O3 ⁱⁱ | 0.96 | 2.19 | 2.749 (3) | 116 |
| O6—H6C \cdots O2 | 0.97 | 1.86 | 2.783 (3) | 160 |
| O7—H7B \cdots O4 ⁱⁱⁱ | 0.96 | 2.25 | 2.770 (3) | 113 |
| O7—H7C \cdots O2 ^{iv} | 0.95 | 1.85 | 2.760 (3) | 159 |

Symmetry codes: (i) $x, y, z - 1$; (ii) $x + 1, y, z$; (iii) $-x + 1, -y, -z + 1$; (iv) $-x + 2, -y, -z + 1$.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Bruker, 2000); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2398).

References

- Bruker (2000). *SADABS*, *SMART*, *SAINT* and *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Tai, X.-S., Yin, X.-H., Tan, M.-Y. & Li, Y.-Z. (2003). *Acta Cryst.* **E59**, o681–o682.

supplementary materials

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Hexaaquamagnesium(II) bis[4-(2-hydroxybenzylideneamino)benzenesulfonate]

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Comment

As part of our ongoing studies of the coordination chemistry of Schiff base ligands (Tai *et al.*, 2003), we now report the synthesis and structure of the title compound, (I), (Fig. 1). Here, however, the ligand is not coordinated to the metal, but instead, a molecular salt arises.

In the crystal of (I), the Mg(II) center (site symmetry $\bar{1}$) is six-coordinate to water molecules. The C7—N1 distance [1.251 (5) Å] in the anion is close to a double-bond value. Otherwise, the geometrical parameters for (I) are normal. The dihedral angle between the two benzene ring is 32.6 (2)°, indicating that the molecule is non-planar, which perhaps correlates with the intramolecular hydrogen bond (Table 1).

Experimental

1 mmol of magnesium perchlorate was added to a solution of salicylaldehyde-4-aminobenzene sulfonic acid (1 mmol) in 10 ml of 95% ethanol. The mixture was stirred for 3 h at refluxing temperature. Evaporating some ethanol, clear blocks of (I) were obtained after two weeks.

Refinement

The O-bound H atoms were located in difference maps and refined as riding in their as-found relative positions with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{O})$.

The other H atoms were placed geometrically (C—H = 0.93–0.96 Å, and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

Figures

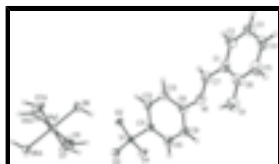


Fig. 1. The molecular structure of (I) showing 30% displacement ellipsoids (arbitrary spheres for the H atoms). Atoms with the suffix A are generated by the symmetry operator ($2 - x, -y, -z$).

Hexaaquamagnesium(II) bis[4-(2-hydroxybenzylideneamino)benzenesulfonate]

Crystal data

[Mg(H₂O)₆](C₁₃H₁₀NO₄S)₂

$M_r = 684.97$

$F_{000} = 716$

$D_x = 1.470 \text{ Mg m}^{-3}$

supplementary materials

Monoclinic, $P2_1/n$

Hall symbol: -P 2yn

$a = 6.2997 (17) \text{ \AA}$

$b = 35.313 (9) \text{ \AA}$

$c = 6.9561 (19) \text{ \AA}$

$\beta = 90.75 (1)^\circ$

$V = 1547.3 (7) \text{ \AA}^3$

$Z = 2$

Mo $K\alpha$ radiation

$\lambda = 0.71073 \text{ \AA}$

Cell parameters from 3237 reflections

$\theta = 2.3\text{--}26.9^\circ$

$\mu = 0.26 \text{ mm}^{-1}$

$T = 291 (2) \text{ K}$

Block, brown

$0.30 \times 0.24 \times 0.22 \text{ mm}$

Data collection

Bruker SMART APEX CCD
diffractometer

Radiation source: sealed tube

Monochromator: graphite

$T = 291(2) \text{ K}$

phi and ω scans

Absorption correction: multi-scan
(SADABS; Bruker, 2000)

$T_{\min} = 0.93$, $T_{\max} = 0.94$

8390 measured reflections

3041 independent reflections

2173 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.044$

$\theta_{\max} = 26.0^\circ$

$\theta_{\min} = 3.0^\circ$

$h = -7 \rightarrow 7$

$k = -22 \rightarrow 43$

$l = -8 \rightarrow 8$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.064$

$wR(F^2) = 0.146$

$S = 1.00$

3041 reflections

205 parameters

Primary atom site location: structure-invariant direct
methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring
sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0657P)^2 + 0.77P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.25 \text{ e \AA}^{-3}$

$\Delta\rho_{\min} = -0.46 \text{ e \AA}^{-3}$

Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|--------------|--------------|--------------|----------------------------------|
| C1 | 0.5147 (7) | 0.29797 (12) | 0.5190 (6) | 0.0540 (10) |
| C2 | 0.6701 (8) | 0.32508 (13) | 0.5548 (6) | 0.0647 (12) |
| H2 | 0.8064 | 0.3173 | 0.5893 | 0.078* |
| C3 | 0.6272 (9) | 0.36279 (14) | 0.5404 (7) | 0.0724 (13) |
| H3 | 0.7326 | 0.3806 | 0.5663 | 0.087* |
| C4 | 0.4255 (9) | 0.37428 (13) | 0.4871 (7) | 0.0712 (14) |
| H4 | 0.3957 | 0.4000 | 0.4748 | 0.085* |
| C5 | 0.2695 (9) | 0.34828 (14) | 0.4524 (6) | 0.0696 (13) |
| H5 | 0.1344 | 0.3564 | 0.4158 | 0.083* |
| C6 | 0.3106 (8) | 0.31045 (12) | 0.4710 (6) | 0.0575 (11) |
| C7 | 0.5634 (7) | 0.25773 (12) | 0.5315 (5) | 0.0532 (10) |
| H7 | 0.7020 | 0.2506 | 0.5614 | 0.064* |
| C8 | 0.4826 (6) | 0.19365 (11) | 0.5055 (5) | 0.0466 (9) |
| C9 | 0.3252 (6) | 0.16787 (12) | 0.5564 (6) | 0.0556 (10) |
| H9 | 0.1914 | 0.1765 | 0.5901 | 0.067* |
| C10 | 0.3670 (6) | 0.13009 (12) | 0.5569 (6) | 0.0495 (9) |
| H10 | 0.2622 | 0.1130 | 0.5925 | 0.059* |
| C11 | 0.5665 (5) | 0.11689 (10) | 0.5043 (4) | 0.0349 (7) |
| C12 | 0.7255 (6) | 0.14225 (11) | 0.4507 (5) | 0.0455 (9) |
| H12 | 0.8588 | 0.1336 | 0.4155 | 0.055* |
| C13 | 0.6805 (6) | 0.18019 (12) | 0.4514 (6) | 0.0525 (10) |
| H13 | 0.7845 | 0.1973 | 0.4148 | 0.063* |
| Mg1 | 1.0000 | 0.0000 | 0.0000 | 0.0385 (4) |
| N1 | 0.4280 (5) | 0.23230 (10) | 0.5038 (5) | 0.0536 (8) |
| O1 | 0.1544 (5) | 0.28509 (10) | 0.4396 (5) | 0.0771 (10) |
| H1C | 0.2119 | 0.2635 | 0.4541 | 0.092* |
| O2 | 0.8439 (3) | 0.06285 (7) | 0.4953 (3) | 0.0406 (6) |
| O3 | 0.5166 (4) | 0.05345 (7) | 0.3242 (3) | 0.0465 (6) |
| O4 | 0.5167 (4) | 0.05225 (7) | 0.6716 (3) | 0.0452 (6) |
| O5 | 0.7139 (3) | 0.02654 (7) | −0.0016 (3) | 0.0421 (6) |
| H5A | 0.7144 | 0.0460 | −0.0981 | 0.063* |
| H5B | 0.6887 | 0.0377 | 0.1221 | 0.063* |
| O6 | 1.1087 (4) | 0.03733 (8) | 0.2055 (4) | 0.0536 (7) |
| H6B | 1.2111 | 0.0247 | 0.2852 | 0.080* |
| H6C | 0.9918 | 0.0457 | 0.2832 | 0.080* |
| O7 | 0.8927 (4) | −0.03449 (8) | 0.2191 (4) | 0.0519 (7) |
| H7B | 0.7912 | −0.0209 | 0.2956 | 0.078* |
| H7C | 1.0104 | −0.0419 | 0.2983 | 0.078* |
| S1 | 0.61427 (12) | 0.06773 (2) | 0.49840 (11) | 0.0325 (2) |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|----|-----------|-----------|-----------|-------------|-------------|-------------|
| C1 | 0.067 (3) | 0.050 (2) | 0.046 (2) | 0.0107 (19) | 0.0090 (19) | 0.0048 (17) |

supplementary materials

| | | | | | | |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| C2 | 0.070 (3) | 0.053 (3) | 0.071 (3) | 0.002 (2) | 0.002 (2) | 0.005 (2) |
| C3 | 0.086 (4) | 0.057 (3) | 0.075 (3) | -0.008 (3) | 0.014 (3) | 0.003 (2) |
| C4 | 0.107 (4) | 0.044 (2) | 0.063 (3) | 0.018 (3) | 0.012 (3) | 0.002 (2) |
| C5 | 0.098 (4) | 0.059 (3) | 0.052 (3) | 0.028 (3) | -0.008 (2) | 0.010 (2) |
| C6 | 0.072 (3) | 0.052 (2) | 0.048 (2) | 0.003 (2) | -0.004 (2) | 0.0087 (18) |
| C7 | 0.057 (3) | 0.052 (2) | 0.050 (2) | 0.015 (2) | -0.0015 (18) | 0.0063 (18) |
| C8 | 0.049 (2) | 0.041 (2) | 0.049 (2) | 0.0062 (17) | -0.0054 (16) | 0.0030 (16) |
| C9 | 0.036 (2) | 0.059 (3) | 0.072 (3) | 0.0088 (18) | 0.0053 (18) | -0.002 (2) |
| C10 | 0.0361 (19) | 0.053 (2) | 0.059 (2) | 0.0043 (17) | 0.0096 (16) | -0.0018 (18) |
| C11 | 0.0269 (16) | 0.0469 (19) | 0.0309 (15) | -0.0006 (13) | -0.0022 (12) | 0.0033 (14) |
| C12 | 0.0388 (19) | 0.047 (2) | 0.051 (2) | 0.0036 (16) | 0.0147 (16) | 0.0013 (16) |
| C13 | 0.051 (2) | 0.049 (2) | 0.058 (2) | -0.0100 (18) | 0.0121 (18) | -0.0013 (18) |
| Mg1 | 0.0372 (9) | 0.0385 (9) | 0.0398 (9) | -0.0013 (7) | -0.0019 (6) | -0.0002 (7) |
| N1 | 0.057 (2) | 0.052 (2) | 0.0507 (19) | 0.0073 (17) | -0.0018 (15) | 0.0017 (15) |
| O1 | 0.069 (2) | 0.066 (2) | 0.096 (3) | 0.0148 (17) | -0.0236 (18) | 0.0034 (17) |
| O2 | 0.0178 (10) | 0.0545 (15) | 0.0494 (13) | 0.0060 (9) | -0.0004 (9) | -0.0018 (11) |
| O3 | 0.0340 (12) | 0.0584 (16) | 0.0470 (14) | 0.0014 (11) | -0.0062 (10) | -0.0143 (12) |
| O4 | 0.0306 (12) | 0.0580 (16) | 0.0471 (14) | 0.0023 (11) | 0.0035 (10) | 0.0138 (12) |
| O5 | 0.0310 (12) | 0.0564 (15) | 0.0388 (12) | 0.0093 (11) | 0.0012 (9) | -0.0032 (11) |
| O6 | 0.0300 (13) | 0.077 (2) | 0.0534 (16) | 0.0003 (12) | -0.0036 (11) | -0.0246 (14) |
| O7 | 0.0286 (12) | 0.0743 (19) | 0.0528 (15) | -0.0006 (11) | 0.0012 (10) | 0.0264 (13) |
| S1 | 0.0202 (4) | 0.0428 (5) | 0.0345 (4) | 0.0020 (3) | -0.0012 (3) | -0.0007 (3) |

Geometric parameters (\AA , $^\circ$)

| | | | |
|----------|-----------|---------------------|-----------|
| C1—C2 | 1.389 (6) | C11—C12 | 1.398 (5) |
| C1—C6 | 1.396 (6) | C11—S1 | 1.762 (4) |
| C1—C7 | 1.456 (6) | C12—C13 | 1.369 (5) |
| C2—C3 | 1.362 (7) | C12—H12 | 0.9300 |
| C2—H2 | 0.9300 | C13—H13 | 0.9300 |
| C3—C4 | 1.380 (7) | Mg1—O5 | 2.031 (2) |
| C3—H3 | 0.9300 | Mg1—O5 ⁱ | 2.031 (2) |
| C4—C5 | 1.364 (7) | Mg1—O6 | 2.055 (2) |
| C4—H4 | 0.9300 | Mg1—O6 ⁱ | 2.055 (2) |
| C5—C6 | 1.366 (6) | Mg1—O7 ⁱ | 2.071 (2) |
| C5—H5 | 0.9300 | Mg1—O7 | 2.071 (2) |
| C6—O1 | 1.346 (5) | O1—H1C | 0.8500 |
| C7—N1 | 1.251 (5) | O2—S1 | 1.457 (2) |
| C7—H7 | 0.9300 | O3—S1 | 1.443 (2) |
| C8—C13 | 1.391 (5) | O4—S1 | 1.465 (2) |
| C8—C9 | 1.395 (6) | O5—H5A | 0.9599 |
| C8—N1 | 1.408 (5) | O5—H5B | 0.9618 |
| C9—C10 | 1.360 (6) | O6—H6B | 0.9551 |
| C9—H9 | 0.9300 | O6—H6C | 0.9653 |
| C10—C11 | 1.394 (5) | O7—H7B | 0.9645 |
| C10—H10 | 0.9300 | O7—H7C | 0.9544 |
| C2—C1—C6 | 118.0 (4) | C12—C13—C8 | 121.4 (4) |
| C2—C1—C7 | 121.0 (4) | C12—C13—H13 | 119.3 |

| | | | |
|-------------|-----------|--------------------------------------|-------------|
| C6—C1—C7 | 121.0 (4) | C8—C13—H13 | 119.3 |
| C3—C2—C1 | 121.5 (5) | O5—Mg1—O5 ⁱ | 180.0 |
| C3—C2—H2 | 119.3 | O5—Mg1—O6 | 89.73 (10) |
| C1—C2—H2 | 119.3 | O5 ⁱ —Mg1—O6 | 90.27 (10) |
| C2—C3—C4 | 119.2 (5) | O5—Mg1—O6 ⁱ | 90.27 (10) |
| C2—C3—H3 | 120.4 | O5 ⁱ —Mg1—O6 ⁱ | 89.73 (10) |
| C4—C3—H3 | 120.4 | O6—Mg1—O6 ⁱ | 180.0 |
| C5—C4—C3 | 120.6 (4) | O5—Mg1—O7 ⁱ | 91.30 (10) |
| C5—C4—H4 | 119.7 | O5 ⁱ —Mg1—O7 ⁱ | 88.70 (10) |
| C3—C4—H4 | 119.7 | O6—Mg1—O7 ⁱ | 91.47 (11) |
| C4—C5—C6 | 120.4 (5) | O6 ⁱ —Mg1—O7 ⁱ | 88.53 (11) |
| C4—C5—H5 | 119.8 | O5—Mg1—O7 | 88.70 (10) |
| C6—C5—H5 | 119.8 | O5 ⁱ —Mg1—O7 | 91.30 (10) |
| O1—C6—C5 | 119.9 (4) | O6—Mg1—O7 | 88.53 (11) |
| O1—C6—C1 | 119.9 (4) | O6 ⁱ —Mg1—O7 | 91.47 (11) |
| C5—C6—C1 | 120.2 (5) | O7 ⁱ —Mg1—O7 | 180.0 |
| N1—C7—C1 | 123.3 (4) | C7—N1—C8 | 121.9 (4) |
| N1—C7—H7 | 118.4 | C6—O1—H1C | 105.6 |
| C1—C7—H7 | 118.4 | Mg1—O5—H5A | 108.8 |
| C13—C8—C9 | 119.2 (4) | Mg1—O5—H5B | 110.0 |
| C13—C8—N1 | 123.3 (4) | H5A—O5—H5B | 109.5 |
| C9—C8—N1 | 117.4 (4) | Mg1—O6—H6B | 108.7 |
| C10—C9—C8 | 120.2 (4) | Mg1—O6—H6C | 109.7 |
| C10—C9—H9 | 119.9 | H6B—O6—H6C | 109.4 |
| C8—C9—H9 | 119.9 | Mg1—O7—H7B | 109.8 |
| C9—C10—C11 | 120.2 (4) | Mg1—O7—H7C | 109.1 |
| C9—C10—H10 | 119.9 | H7B—O7—H7C | 109.5 |
| C11—C10—H10 | 119.9 | O3—S1—O2 | 111.04 (14) |
| C10—C11—C12 | 120.5 (3) | O3—S1—O4 | 112.40 (15) |
| C10—C11—S1 | 119.4 (3) | O2—S1—O4 | 113.27 (14) |
| C12—C11—S1 | 120.1 (3) | O3—S1—C11 | 107.02 (15) |
| C13—C12—C11 | 118.5 (3) | O2—S1—C11 | 106.65 (14) |
| C13—C12—H12 | 120.8 | O4—S1—C11 | 105.96 (15) |
| C11—C12—H12 | 120.8 | | |

Symmetry codes: (i) $-x+2, -y, -z$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

| <i>D</i> —H \cdots <i>A</i> | <i>D</i> —H | H \cdots <i>A</i> | <i>D</i> \cdots <i>A</i> | <i>D</i> —H \cdots <i>A</i> |
|-----------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| O1—H1C \cdots N1 | 0.85 | 1.78 | 2.574 (5) | 154 |
| O5—H5A \cdots O4 ⁱⁱ | 0.96 | 2.03 | 2.732 (3) | 129 |
| O5—H5B \cdots O3 | 0.96 | 1.87 | 2.768 (3) | 154 |
| O6—H6B \cdots O3 ⁱⁱⁱ | 0.96 | 2.19 | 2.749 (3) | 116 |
| O6—H6C \cdots O2 | 0.97 | 1.86 | 2.783 (3) | 160 |
| O7—H7B \cdots O4 ^{iv} | 0.96 | 2.25 | 2.770 (3) | 113 |
| O7—H7C \cdots O2 ^v | 0.95 | 1.85 | 2.760 (3) | 159 |

supplementary materials

Symmetry codes: (ii) $x, y, z-1$; (iii) $x+1, y, z$; (iv) $-x+1, -y, -z+1$; (v) $-x+2, -y, -z+1$.

Fig. 1

